

NARINGENIN COUMAROYLGLUCOSIDES FROM *MABEA CAUDATA**

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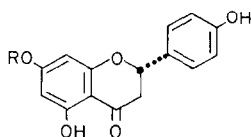
(Received 1 December 1981)

Key Word Index—*Mabea caudata*; Euphorbiaceae; naringenin; naringenin 7-*O*-(*p*-coumaroylglucosides).

Abstract—The fruits of *Mabea caudata* contain, besides 5,7,4'-trihydroxyflavanone (naringenin), naringenin 7-*O*-β-(3,6-di-*p*-coumaroylglucoside) and naringenin 7-*O*-β-(3-*p*-coumaroylglucoside).

INTRODUCTION

The genus *Mabea* Aubl. comprises 50 species distributed over tropical Central and South America [2]. One of these species, *M. caudata* Peth., is well represented in Amazonian forest regions. Fruits of the tree yielded (2*S*)-5,7,4'-flavanone (naringenin, **1a**) and two novel compounds (**1b**, **1c**).



1a R = H

1b R = β-3, 6-Di-*p*-Coumaroylglucosyl

1c R = β-3-*p*-Coumaroylglucosyl

RESULTS AND DISCUSSION

Acid hydrolysis of **1b** and **1c** gave, besides glucose and *p*-coumaric acid, naringenin. In both compounds the C-7 hydroxyl group of the naringenin moiety must be substituted, since, as with the aglycone, UV AlCl_3 shifts are observed, but, unlike the aglycone, the UV spectra are not changed upon addition of NaOAc. All 270-MHz ^1H NMR spectral features of **1b** are amenable to first-order analysis (Table 1) and led to assignments which were confirmed by the complete series of decoupling experiments. The signals due to H-3 (δ 5.28) and both H-6 (δ 4.37, 4.60) of the glucose moiety appear at relatively low field demonstrating esterification of the hydroxyls at these sites by *p*-coumaric acid. The analysis of the analogous spec-

trum of **1c** is less straightforward, only H-1 and H-2 giving fully resolved signals (respectively at δ 5.26 and 3.70). Here only one of the carbinolic protons of the glucose moiety is represented by a signal at relatively low field (δ 5.23). This must belong to H-3 since double irradiation at this frequency affected the H-2 signal and vice versa.

With respect to stereochemical features, the protons on the anomeric carbons in both, **1b** and **1c**, occupy α -positions, in view of their axial-axial relationships with H-2 ($J = 8$ Hz, Table 1). The 2*S*-configurational assignment for naringenin (**1a**) results from the observation of a positive Cotton effect for the $\pi \rightarrow \pi^*$ transition [3].

EXPERIMENTAL

Isolation of the constituents. Fruits of *M. caudata* were collected by Hipólito F. Paulino Filho near Humaitá, Amazonas State, from a specimen identified by Dr. William A. Rodrigues, INPA, Manaus. The powder (250 g), obtained from dry fruits, was percolated in succession with C_6H_6 and EtOH. The EtOH extract (6.5 g) was suspended in Me_2CO and filtered. The Me_2CO extract (5 g) was chromatographed on a dry column (100 g Si gel deactivated with 10% H_2O , C_6H_6 -EtOAc, 1:1). The extruded column was cut into 6 equal segments numbered 1–6 from column bottom to top. The segments were eluted with Me_2CO . Eluate 1 (120 mg) was washed with CHCl_3 and purified by prep. TLC (Si gel, C_6H_6 - Me_2CO , 7:3) to give **1a** (16 mg). Eluates 2 and 3 (2.5 g), repeatedly chromatographed in the same way, yielded slightly less polar **1b** (375 mg) and slightly more polar **1c** (250 mg).

Naringenin (1a), mp 246–248° (MeOH) (lit. [4] mp 248°). ORD (MeOH, 250): $[\phi]_{300}^{25} + 2000$, $[\phi]_{316}^{25} 0$, $[\phi]_{330}^{25} - 1100$, $[\phi]_{345}^{25} - 340$, $[\phi]_{353}^{25} 0$, $[\phi]_{370}^{25} + 400$.

Naringenin 7-*O*-β-(3,6-di-*p*-coumaroylglucose (1b), mp 155–158° (CCl_4 - Me_2CO 8:2). [Found: C, 64.91; H, 4.85. $\text{C}_{39}\text{H}_{34}\text{O}_{14}$ requires: C, 64.46; H, 4.72%.] UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm: 285, 310 (ϵ 50 000, 40 750); no NaOAc shift; $\lambda_{\text{max}}^{\text{MeOH}+\text{AlCl}_3}$ nm: 305 (ϵ 50 300). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3509, 1712, 1642, 1616, 1517, 1443, 1250 (*br*), 837. **Acetate (1b)**, Ac_2O , $\text{C}_5\text{H}_5\text{N}$, 24 hr, room temp.), mp 116–118° (CCl_4 - Me_2CO , 8:2). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 1795, 1629, 1529, 1462, 1397, 1328, 1200 (*br*), 923, 853.

*Part 4 in the series "The Chemistry of Brazilian Euphorbiaceae". For part 3 see ref. [1]. Based on part of the M.Sc. thesis presented by D.A.D.B., on leave of absence from Faculdade de Farmácia e Odontologia de Riberirão Preto, to Universidade de São Paulo (1981).

Table 1. ^1H NMR data of **1a**, **1b**, **1c** and derivatives [270 MHz (**1b**, **1c**) and 60 MHz (**1a** and acetates)], in $\text{Me}_2\text{CO}-d_6$ (**1a**–**1c**) or CDCl_3 acetates, TMS an int. standard]*

Assignments	1a	Acetate of 1a	1b	Acetate of 1b	1c	Acetate of 1c
Naringenin						
H-2	5.46 <i>dd</i> (12, 4)	5.60 <i>dd</i> (14, 4)	5.40 <i>dd</i> (12.5, 3)	5.2–5.4 <i>m</i>	5.50 <i>dd</i> (12.5, 3)	5.2–5.4
H-3	2.66 <i>dd</i> (14, 4)	2.6–3.3 <i>m</i>	2.77 <i>dd</i> (17, 3)	2.7–3.2 <i>m</i>	2.79 <i>dd</i> (17, 3)	2.7–3.2 <i>m</i>
H-3	3.20 <i>dd</i> (14, 12)		3.21 <i>dd</i> (17, 12.5)		3.25 <i>dd</i> (17, 12.5)	
HO-5	11.8 <i>s</i> (<i>br</i>)		—	11.5 <i>s</i>	—	11.5 <i>s</i>
H-6	5.93 <i>s</i>	6.63 <i>d</i>	6.20 <i>d</i>	6.35 <i>d</i>	6.17 <i>d</i>	6.36 <i>d</i>
H-8		(3)	(2)	(3)	(2.5)	(3)
H-8		6.80 <i>d</i> (3)	6.25 <i>d</i> (2)	6.43 <i>d</i> (3)	6.21 <i>d</i> (2.5)	6.43 <i>d</i> (3)
H-2', H-6'	7.35 <i>d</i> (8.5)	7.51 <i>d</i> (9)	7.51 <i>d</i> (8.5)	7–7.5 <i>m</i>	7.40 <i>d</i> (8.5)	7.5 <i>d</i> (8)
H-3', H-5'	6.80 <i>d</i> (8.5)	7.20 <i>d</i> (9)	6.90 <i>d</i> (9)		6.90 <i>d</i> (9)	7.2 <i>d</i> (8)
AcO-5	—	2.36 <i>s</i>	—	2.35 <i>s</i>	—	2.36 <i>s</i>
AcO	—	2.30 <i>s</i>	—		—	2.33 <i>s</i>
AcO	—	2.16 <i>s</i>	—		—	—
Glucose						
H-1			5.32 <i>d</i> (7.5)	5.2–5.4 <i>m</i>	5.26 <i>dd</i> (7.5, 4.5)	5.2–5.4 <i>m</i>
H-2			3.78 <i>dd</i> (9, 7.5)		3.70 <i>dd</i> (9.5, 7.5)	
H-3			5.28 <i>t</i> (9)		5.23 <i>m</i>	
H-4			3.78 <i>dd</i> (9, 7.5)		3.92 <i>~ d</i> (7.5)	
H-5			4.08 <i>ddd</i> (9.5, 6, 2)			
H-6			4.37 <i>dd</i> (12, 2)	4.3–4.6 <i>m</i>	3.75 <i>m</i>	4–4.3 <i>m</i>
H-6			4.60 <i>dd</i> (12, 6)			
AcO			—	2.03 <i>s</i>	—	2.03 <i>s</i>
AcO			—		—	
AcO			—		—	
<i>p</i> -Coumarate						
H- α			6.38 <i>d</i> (16)	6.3 <i>d</i> (16)	6.38 <i>d</i> (16)	6.33 <i>d</i> (16)
			6.40 <i>d</i> (16)		—	—
H- β			7.62 <i>d</i> (16)	7.7 <i>d</i> (16)	7.65 <i>d</i> (16)	7.70 <i>d</i> (16)
			7.67 <i>d</i> (16)		—	—

Table 1—(continued)

Assignments	1a	Acetate of 1a	1b	Acetate of 1b	1c	Acetate of 1c
H-2, H-6			7.37 <i>d</i> (8.5) 7.56 <i>d</i> (8.5)	7.75 <i>m</i>	7.55 <i>d</i> (8.5) —	7.6 <i>d</i> (8) —
H-3, H-5			6.90 <i>d</i> (8.5) 6.90 <i>d</i> (8.5)		6.90 <i>d</i> (8.5) —	7.2 <i>d</i> (8) —
AcO			—		—	2.33 <i>s</i>
			—		—	—

*Coupling constants (Hz) in brackets.

Naringenin 7-O-β-(3-p-coumaroylglucose) (1c), mp 162–164° (EtOH) [Found: C, 62.56; H, 5.01. C₃₀H₂₈O₁₂ requires: C, 62.07; H, 4.86%.] UV λ_{max}^{MeOH} nm: 285, 310 (ε 26 400, 18 200); no NaOAc shift; λ_{max}^{MeOH+AlCl₃} nm: 305 (ε 32 300). IR ν_{max}^{KBr} cm⁻¹: 3450, 1701, 1650, 1618, 1517, 1449, 1250 (*br*), 832. *Acetate (1c, Ac₂O, C₅H₅N, 24 hr, room temp.)*, mp 138–141° (CCl₄). IR ν_{max}^{KBr} cm⁻¹: 1761, 1695, 1623, 1570, 1515, 1449, 1200 (*br*), 917, 845.

Hydrolysis of 1b and 1c. A suspension of the compound (80 mg) in H₂O–conc. HCl, 9:1 (7 ml) was heated under reflux for 4 hr [5], cooled and extracted with EtOAc. The presence of glucose was demonstrated in the aq. soln by direct TLC (Si gel, *n*-C₅H₁₁OH–AcOH–H₂O, 4:1:5) comparison with an authentic sample. The EtOAc extract (*ca* 45 mg) was separated by prep. TLC (Si gel, C₆H₆–Me₂CO, 7:3) into less polar **1a** and more polar *o*-coumaric acid. Both compounds were identified by direct comparison with authentic samples.

Acknowledgements—This work was supported by a FAPESP graduate fellowship to D.A.D.B. and by grants from Conselho Nacional de Desenvolvimento Científico e Tecnológico, Financiadora de Estudos e Projetos and Fundação de Amparo à Pesquisa do Estado de São Paulo.

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